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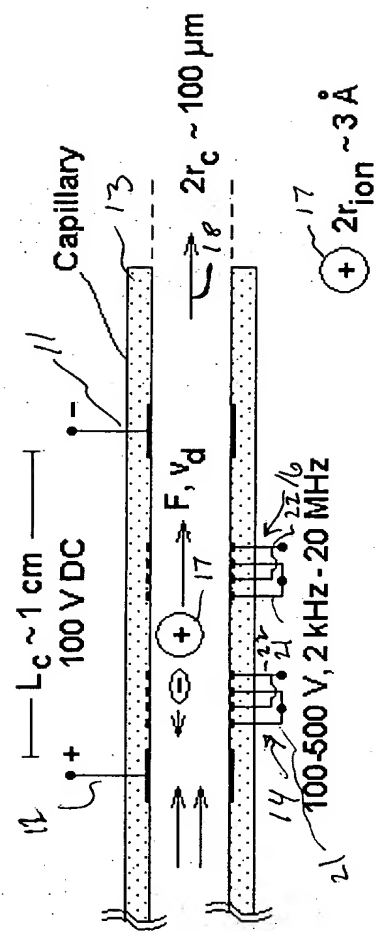


FIGURE 1

1100.1244101

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Applied Pot.		Gas Velocities		Re(Ls)		Δp(Ls)		Qohmic		Ion+Flow		Qvisc+Qgen	
V	Ion	for Ls=Le	Ls	-	-	psid	mW	mW	mW	Qvisc	Qgen	Qtotal	Qideal
100	mol fraction	cm/s	cm/s							mW	mW	mW	mW
100	1.000E-12	0.01	0.00	0.000	0.000	0.000010	0.000000	1.238E-12	0.000001	1.11E-06	1.238E-12	1.238E-12	1.238E-12
	3.162E-12	0.04	0.00	0.000	0.000	0.000031	0.000001	1.238E-11	0.000003	3.51E-06	1.238E-11	1.238E-11	1.238E-11
	1.000E-11	0.12	0.00	0.000	0.000	0.000097	0.000002	1.238E-10	0.000011	1.11E-05	1.238E-10	1.238E-10	1.238E-10
	3.162E-11	0.37	0.01	0.001	0.001	0.000307	0.000005	1.238E-09	0.000035	3.51E-05	1.238E-09	1.238E-09	1.238E-09
	1.000E-10	1.18	0.02	0.002	0.002	0.000971	0.000017	1.238E-08	0.000109	1.11E-04	1.238E-08	1.238E-08	1.238E-08
	3.162E-10	3.72	0.07	0.005	0.005	0.003072	0.000053	1.238E-07	0.000346	3.51E-04	1.238E-07	1.238E-07	1.238E-07
	1.000E-09	11.76	0.24	0.016	0.016	0.009715	0.000168	1.238E-06	0.001095	1.11E-03	1.238E-06	1.238E-06	1.238E-06
	3.162E-09	37.20	0.74	0.050	0.050	0.030721	0.000531	1.238E-05	0.003465	3.53E-03	1.238E-05	1.238E-05	1.238E-05
	1.000E-08	117.62	2.35	0.159	0.159	0.097147	0.001684	0.00012	0.010996	0.01	0.0001	0.0001	0.0001
	3.162E-08	371.96	7.44	0.502	0.502	0.307206	0.005383	0.00124	0.035152	0.04	0.0012	0.0012	0.0012
50	1.000E-07	1176.23	23.52	1.586	1.586	0.971472	0.0017605	0.01238	0.114973	0.13	0.0124	0.0124	0.0124
	3.162E-07	3719.57	74.39	5.016	5.016	3.072064	0.0061508	0.12379	0.401686	0.53	0.1238	0.1238	0.1238
	1.000E-06	11762.30	235.25	15.863	15.863	9.714720	0.0252860	1.23787	1.651345	2.15	1.2379	1.2379	1.2379
	3.162E-06	37195.66	743.91	50.163	50.163	30.720642	0.1383175	12.37871	9.033049	21.55	12.3787	12.3787	12.3787
	1.000E-05	117623.02	2352.46	158.628	158.628	97.147201	1.0209592	123.78713	66.675372	191.48	123.7871	123.7871	123.7871
	Drift Vel. in cm/s, vd =		461.747										
	NA, Avogadro Num. in 1/cm3 =		2.8830E+19										
	q, Electronic Charge in Cb =		1.6022E-19										
	Eion, Ioniz. Energy												
	eV												

Figure 2

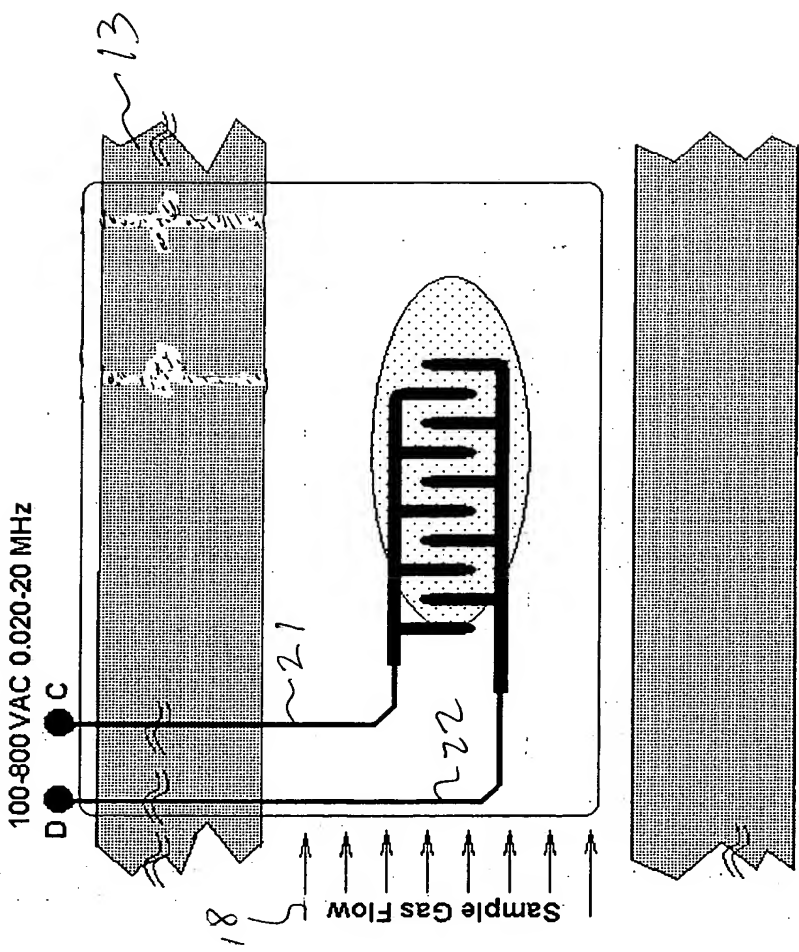


FIGURE 3

# Electron Affinities and Electron Configurations

Element	Electron Affinity (kJ/mol)	Electron Configuration	Ionization Energies (kJ/mole)
H	72.8	1s <sup>1</sup>	1300
He	<0	1s <sup>2</sup>	2400
Li	59.8	[He] 2s <sup>1</sup>	
Be	<0	[He] 2s <sup>2</sup>	
B	27	[He] 2s <sup>2</sup> 2p <sup>1</sup>	
C	122.3	[He] 2s <sup>2</sup> 2p <sup>2</sup>	1050
N	<0	[He] 2s <sup>2</sup> 2p <sup>3</sup>	
O	141.1	[He] 2s <sup>2</sup> 2p <sup>4</sup>	1300
F	328.0	[He] 2s <sup>2</sup> 2p <sup>5</sup>	
Ne	<0	[He] 2s <sup>2</sup> 2p <sup>6</sup>	

Figure 4

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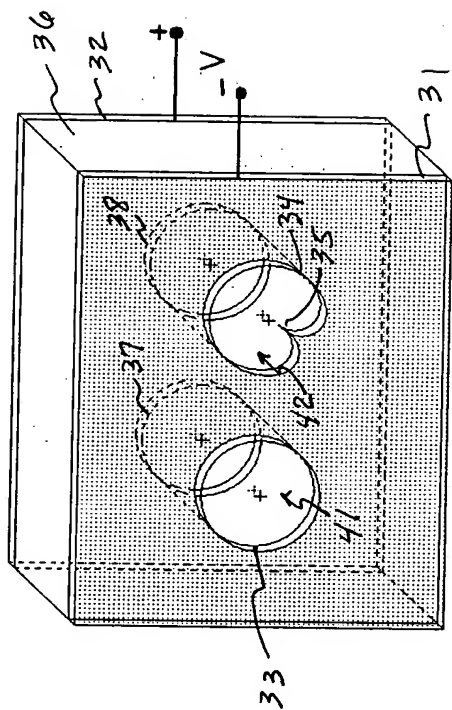


FIGURE 5

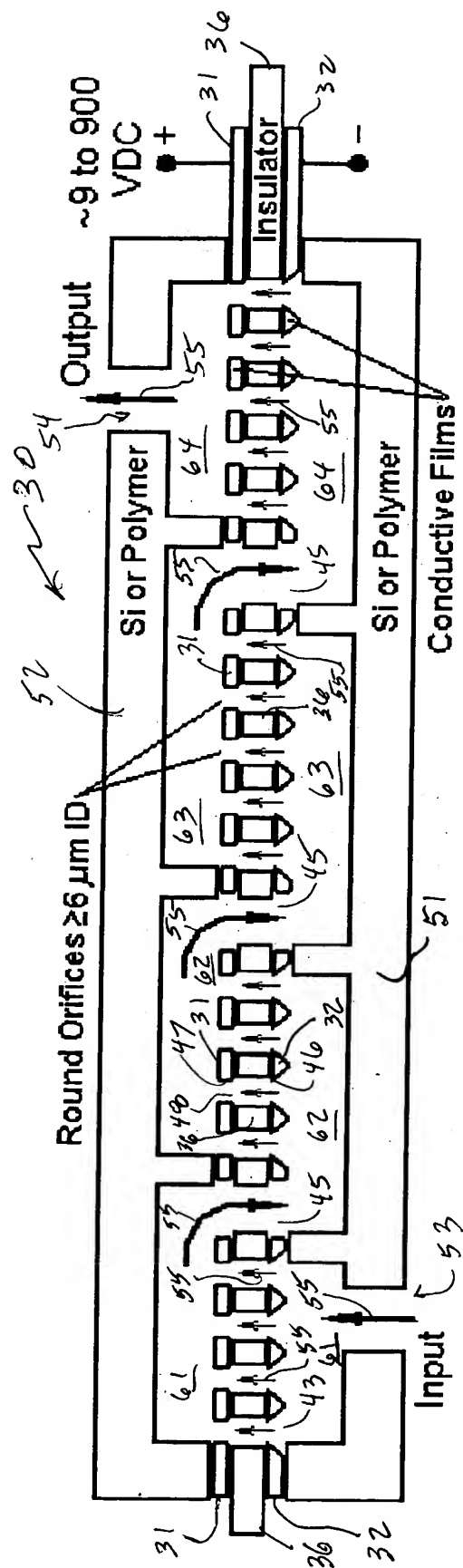


FIGURE 6

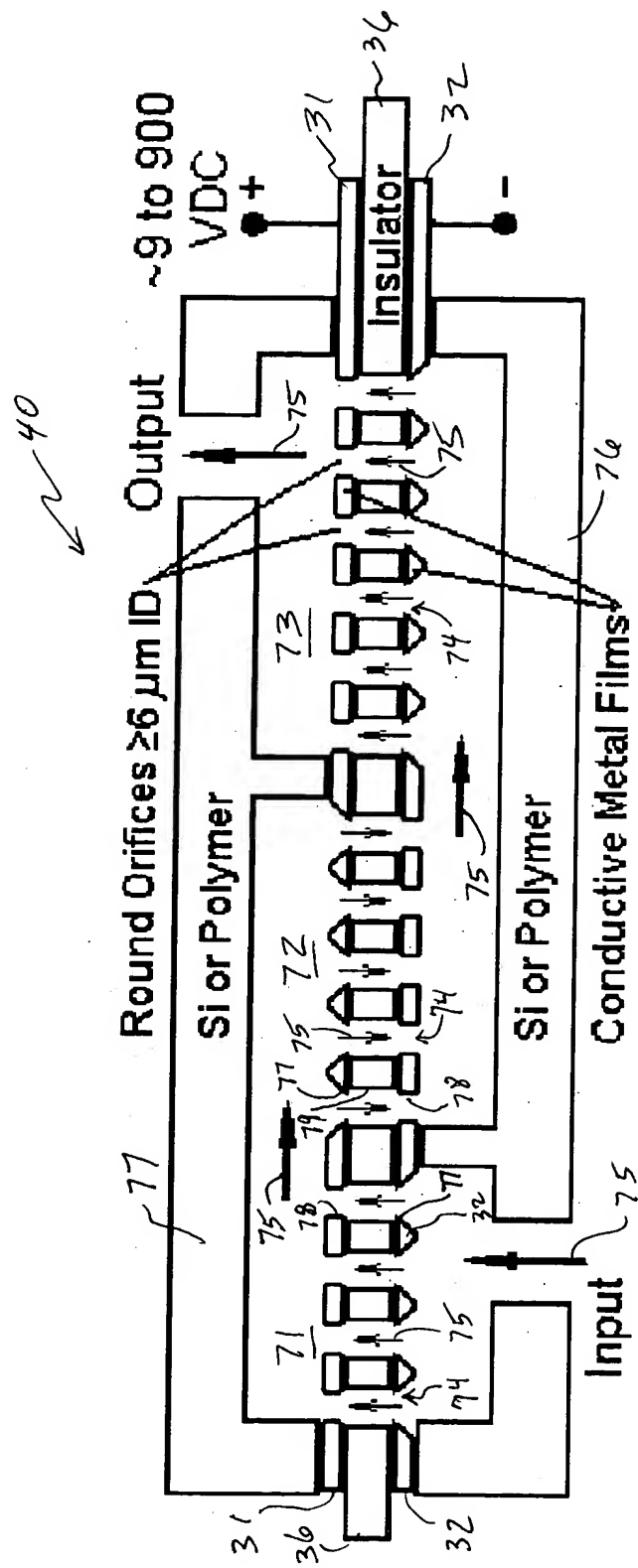


FIGURE 7



Comparison of Performance Between Pumps Based on Different Technologies

Method	Base Unit Size x N mm3	Frequency Hz	Power mW	Voltage V	Flow Rate cm3/min	$\Delta p$ psid
Theoretical Ion Drag	- 10 x 0.25 x 1	= 2.5				
		DC	1.26	1.41	9.7	
		DC	1.65	401.41	9.7	
MesoPump (el.-static, future)	5x5x0.5x15	= 188	14	100	1.0	10
MesoPump (el.-static, today)	10x10x1x50	= 5000	25	150	1.0	10
MesoPump (el.-static, today)	10x10x1x50	= 5000	25	150	1.0	10
Piezo-Electric (Fraunhofer)	7x7x1.1x7x14	= 5282		98		

Figure 8

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Temperature Dependence of Ion Concentration			
Temper. T in K	Ion Aff. Energy E(-) in J/mol	Ionizat. Energy E(+) in J/mol	
	100,000	1,000,000	
	exp(-E/RT)	exp(-E/RT)	
600	2.239E-09	3.169E-87	
1500	3.468E-04	2.514E-35	
2000	2.541E-03	1.123E-26	
2500	8.395E-03	1.739E-21	

FIGURE 9

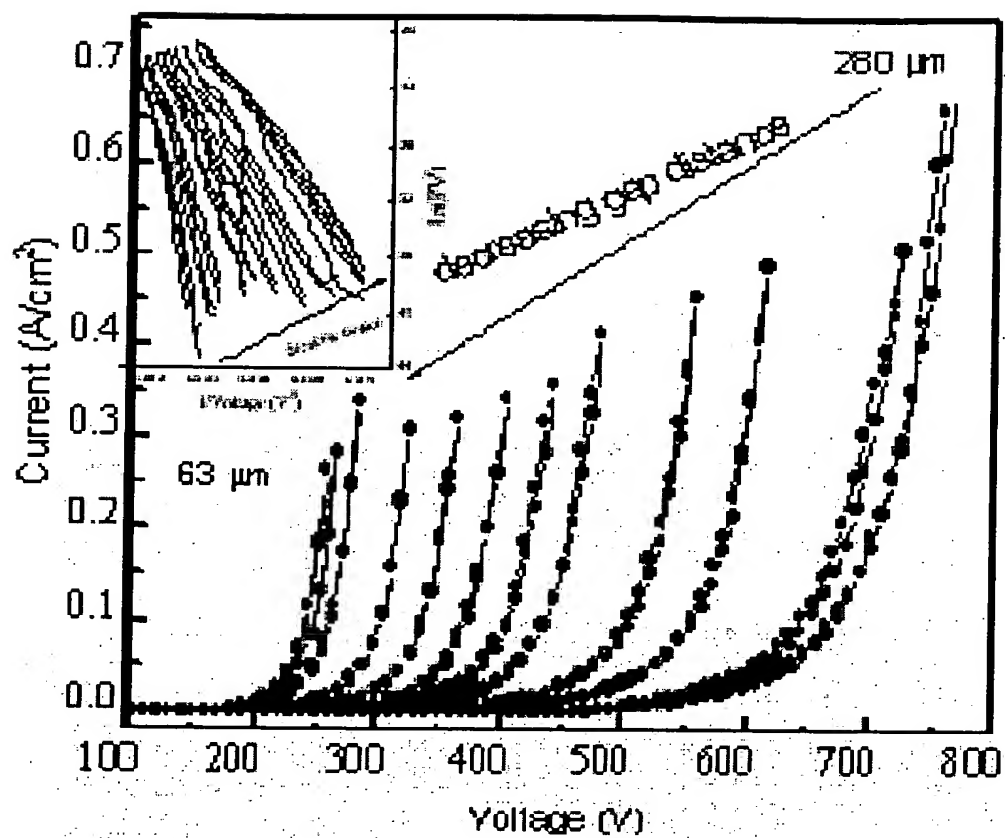


FIGURE 10

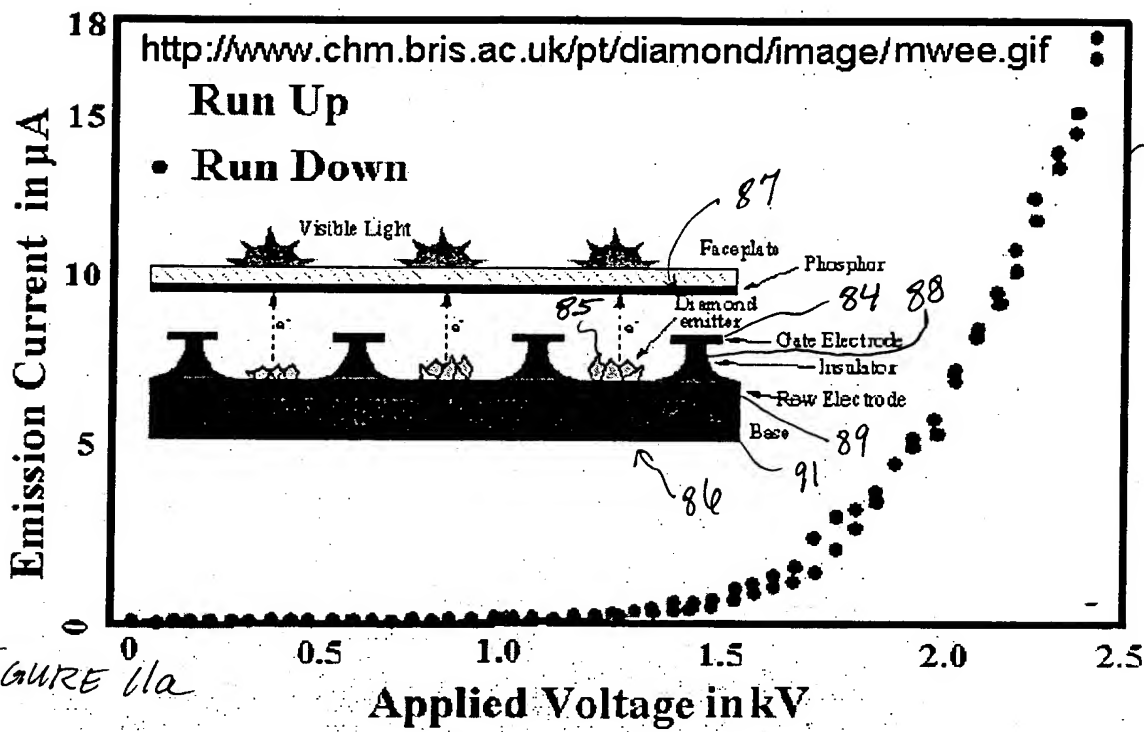


FIGURE 11a

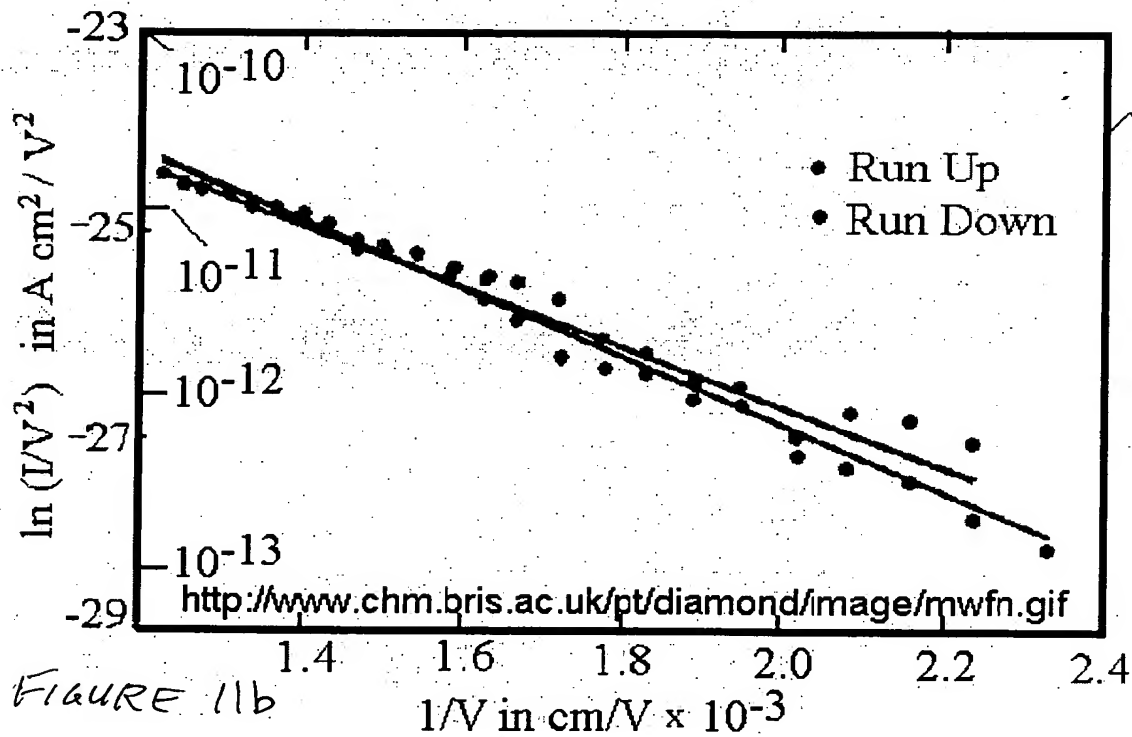


FIGURE 11b